



Engineering the Microstructure of Semicrystalline Polymers

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Semicrystalline Polymers: Who Cares?



Kobe, Japan (January 1995)

6000 dead

500000 homes destroyed

Osaka gas:

Iron/steel pipes failed

NO failures in PE pipes

Entire Japanese pipe market (and most of the European and North American markets) for water and natural gas now use polyethylene piping





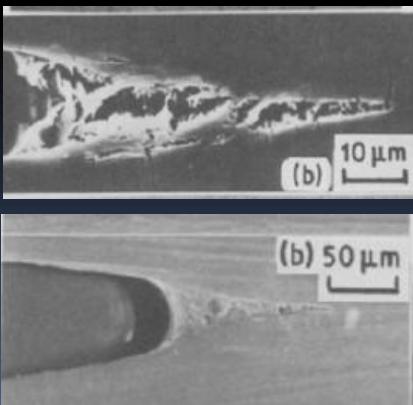
There is polyethylene, and then there is polyethylene...

Semicrystalline polymers ~ 70% of all synthetic polymers
(~225 billion kg/year)

Polyethylene: Example of *semicrystalline* polymer, viz. can crystallize into 3D ordered structure

Microstructure (and properties) can be tailored for same molecule

How does “pipe grade” PE differ from “plastic bucket” PE?



The *microstructure* of pipe grade PE inhibits formation, propagation of cracks

What is *microstructure*, and what controls it?

Source: Lu (1988)

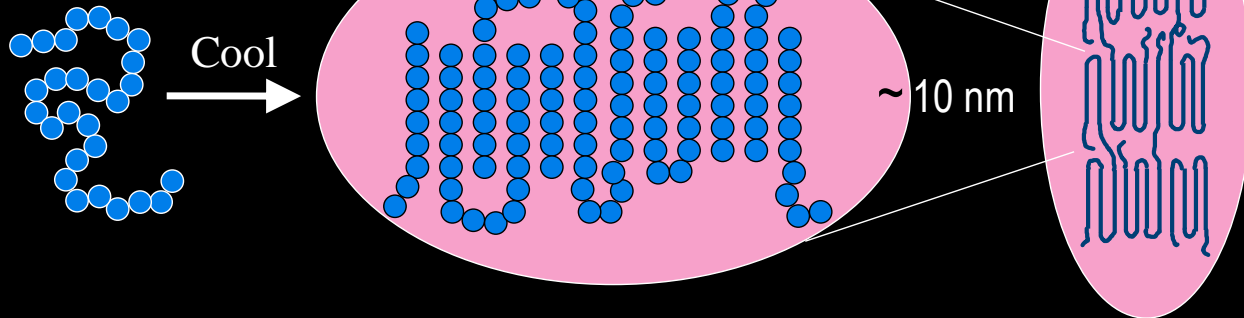
J. Mat. Sci. 1988, 23, 643.



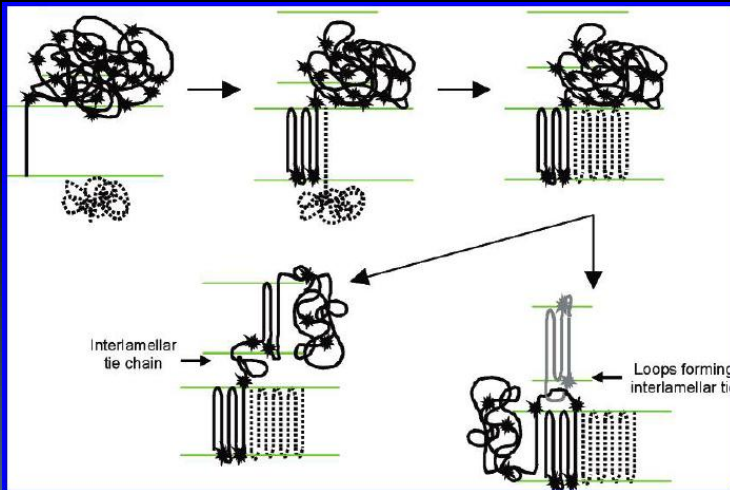
Semicrystalline microstructure

Polymers: Chain-like, form lamellar crystals

High temperature
“melt”



Semi-crystalline: Amorphous and Crystal states co-exist
Far from equilibrium: Gibbs phase rule doesn't hold



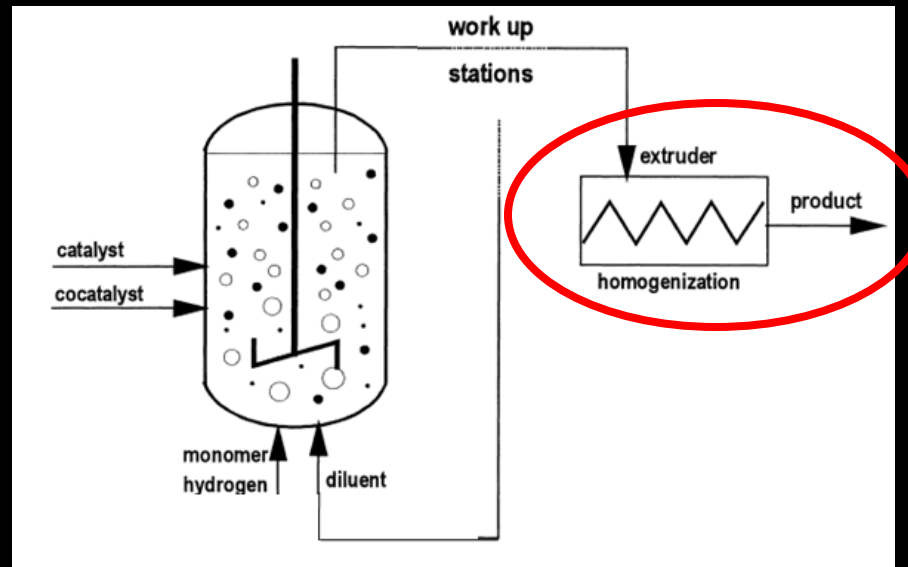
In pipe grade PE

Macro-scale toughness controlled by
Molecular attributes (molecular
weight distribution, chemical
composition along polymer chain)

More “tie” chains that link crystalline
lamellae



Controlling semicrystalline microstructure



Changing the chemistry of the polymer: Changes at the reactor level (difficult, expensive): pipe grade PE

Polymer crystallization is non-equilibrium (exploit this!)

- > Process the polymer differently
- > Engineer desired microstructure
- > Control properties

Challenge: Tools to characterize microstructural development
(Still no way of *measuring* tie chain content)



Controlling semicrystalline microstructure

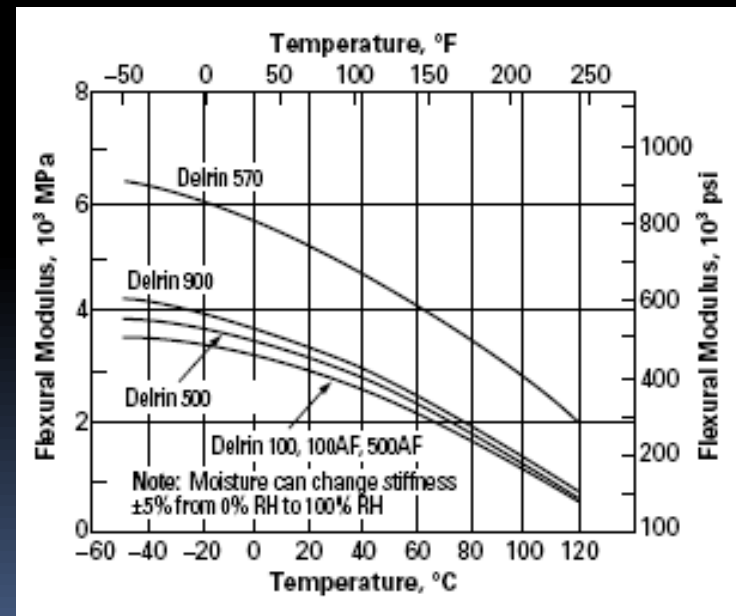


Engineering semicrystalline polymer

Excellent fatigue properties

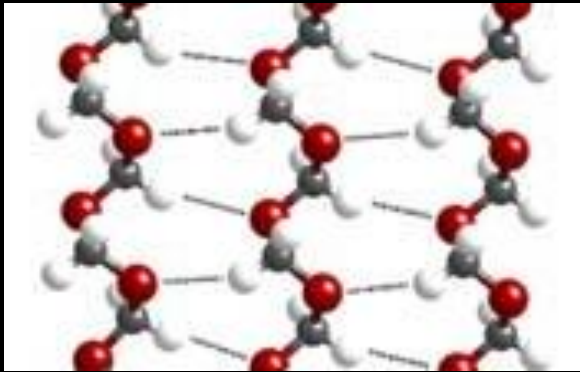
Melts around 175°C, but stiffness deteriorates near 100°C

Why?? What can we do about this?





Temperature dependent change in stiffness

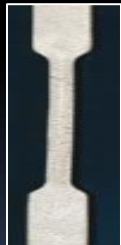


Acta Cryst. 2007, 63, 190.

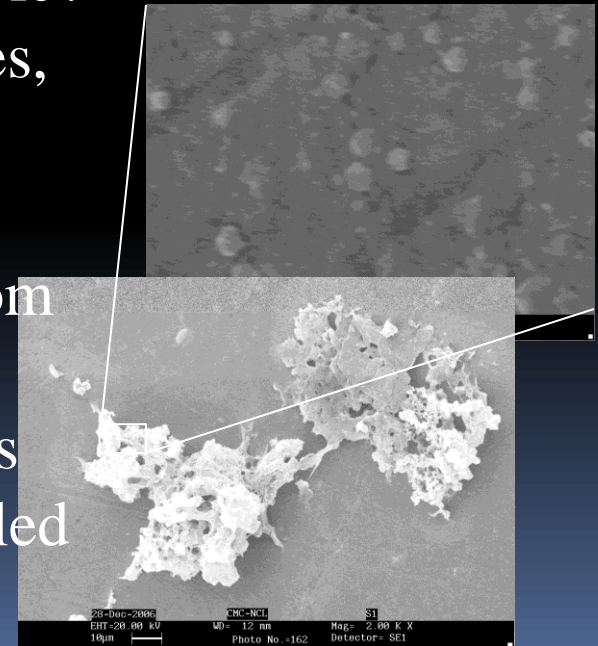
Polyoxymethylene forms helical structures that pack into crystals
Believed that decreased stiffness on heating due to increased torsional motions of chains (intrinsic to the chain chemistry)

Does the interphase between lamellae play a role?
Compare model semicrystalline microstructures,
with same local crystal structure

Injection molded
from the melt.
Chain entanglements
create interphase
structure



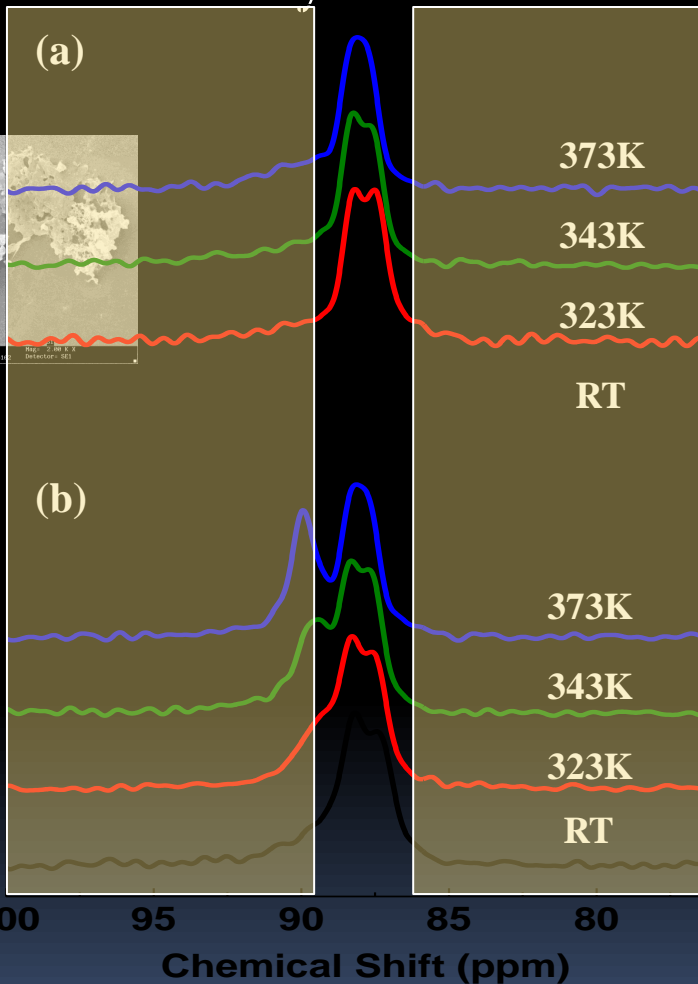
Crystallized from
dilute solution.
Isolated crystals
from disentangled
chains



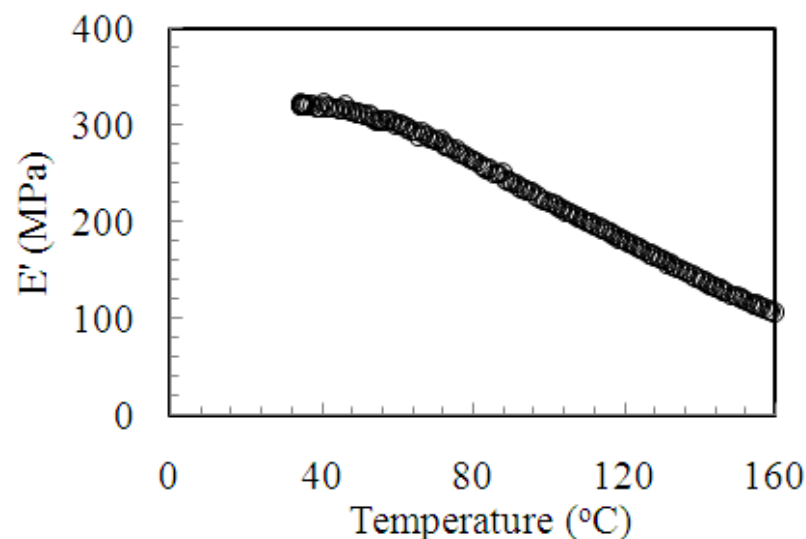
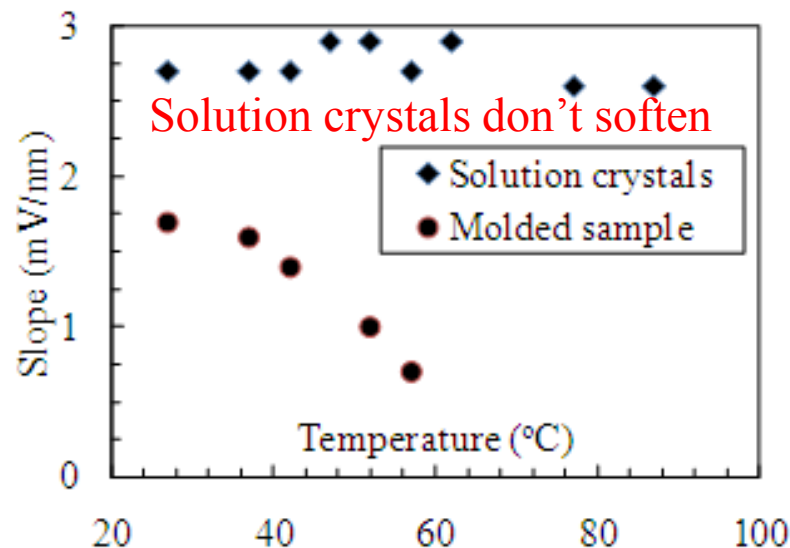


Solid State NMR: Chain mobility in both microstructures

Solution crystals



Melt crystallized

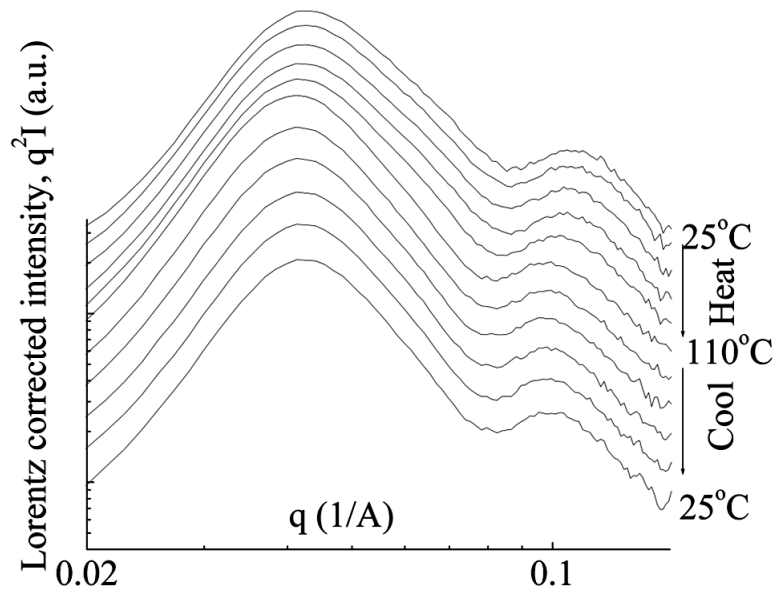
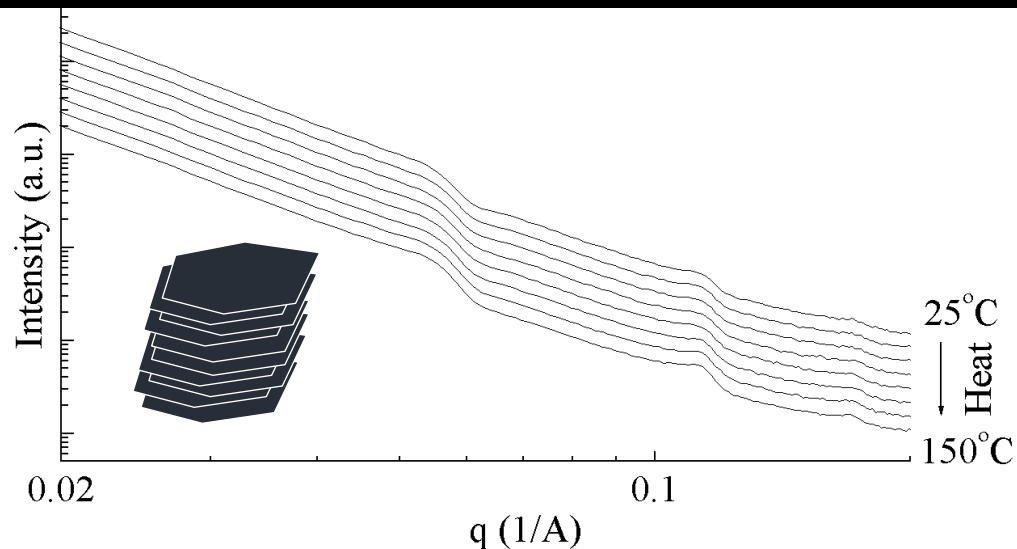




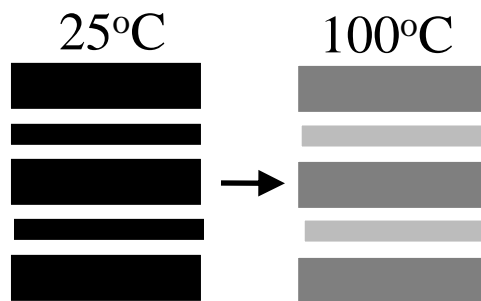
Microstructural changes on heating POM

Use Small Angle X-ray Scattering (SAXS) to probe lamellae

Solution crystals are stacks of ~11 nm crystals
SAXS does not change on heating to 150°C



Melt crystals: Polydisperse lamellae stacked together

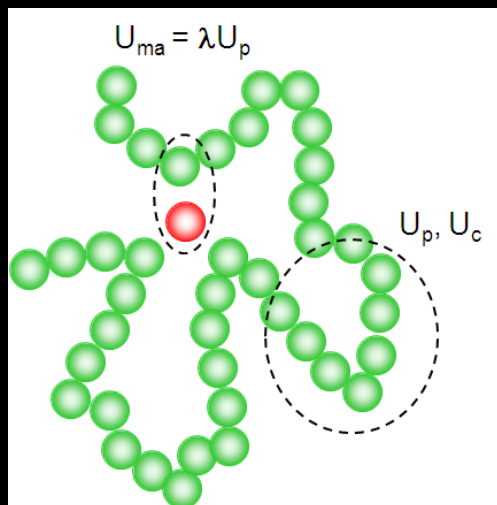


• On heating to 100°C, interspersed thin lamellae melt

(US patent filed
Manuscript in preparation)

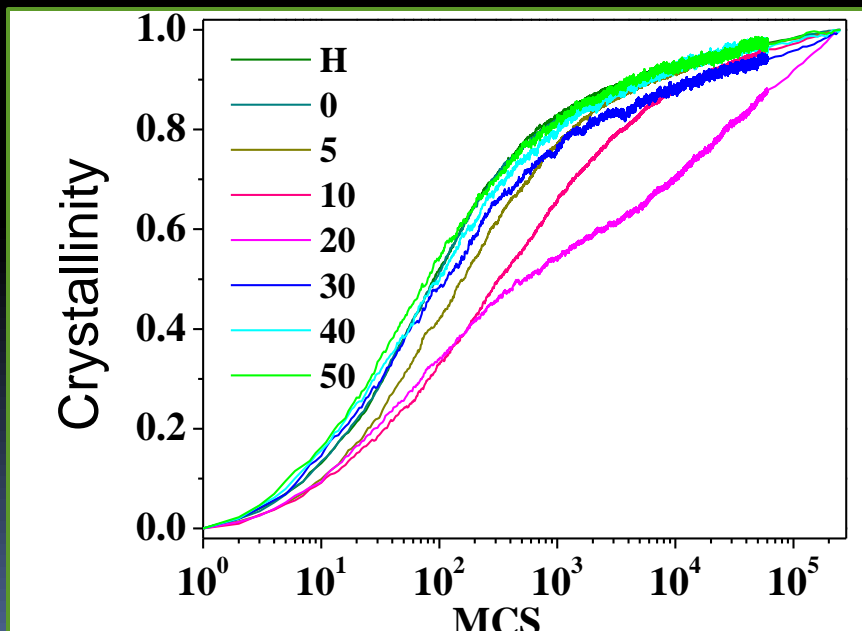
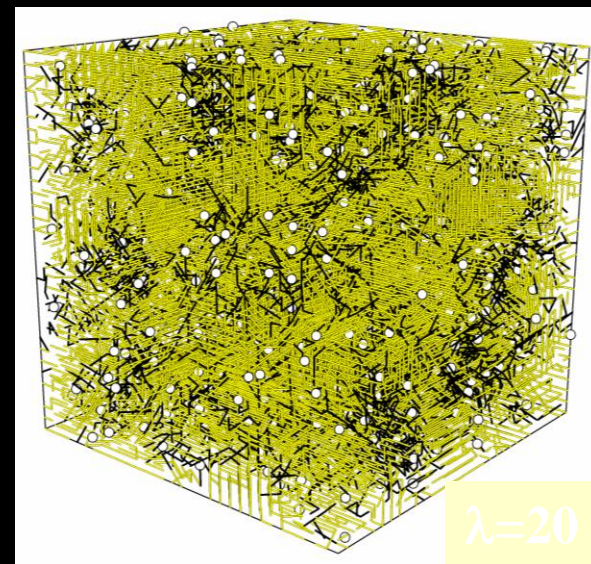


How can we dial-in lamellar structure?



“Sticky” additives that make chains sluggish as they crystallize on cooling

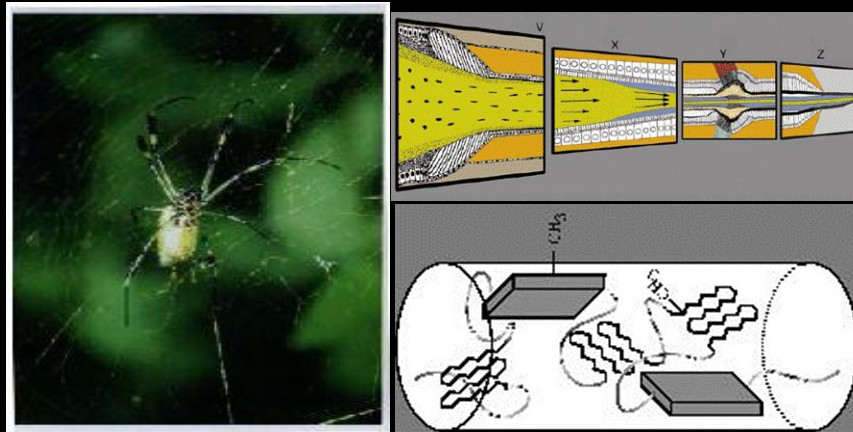
Lattice MC simulations to model crystallization



Suitable choice of additives (with intermediate stickiness) can qualitatively change crystallization.



Semicrystalline μ -structure > Orientation (by flow)



Nature 2001, 410, 541; *Science* 1996, 271, 39

High modulus and toughness

Low viscosity protein “dope”
solution spun by the spider through
a “hyperbolic die”

SPIDER VERSUS SYNTHETIC

Material	Strength (N/m ²)	Energy to break (J/kg)
Dragline silk	1×10^9	1×10^5
Kevlar	4×10^9	3×10^4

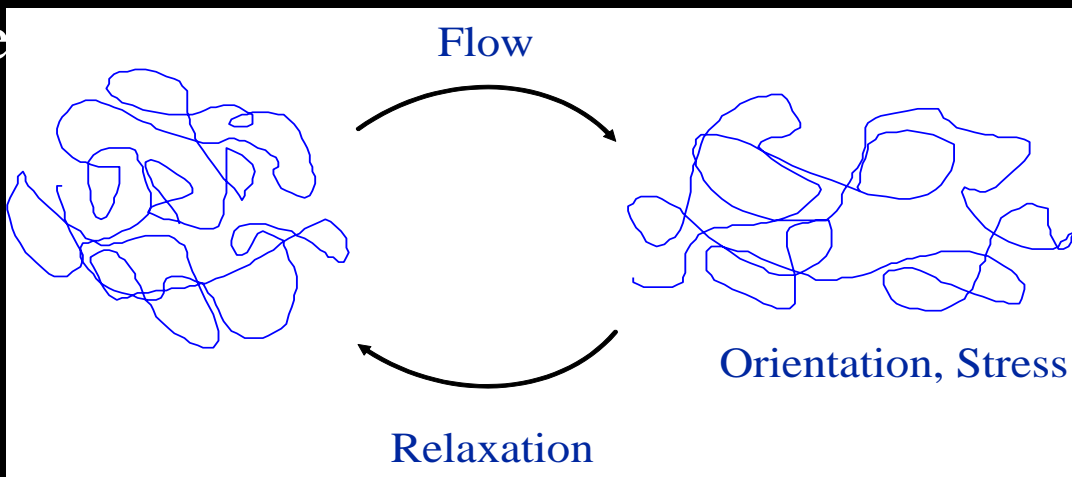
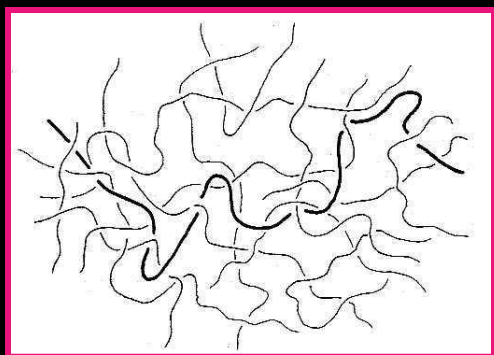
Strong elongational flow induces
orientation



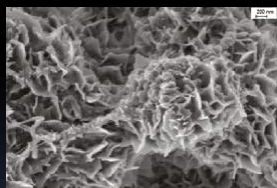
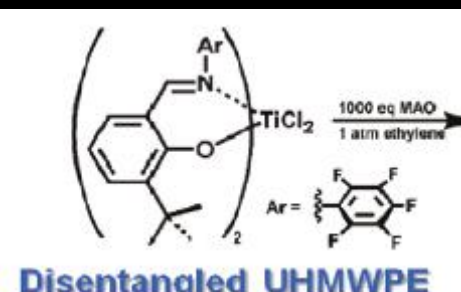
What does flow do?

Polymers are liquid-like
but relax slowly

$\tau_{\text{relax}} \approx 0$ (1-10 ms) at 150°C
for PE with MW = 10^5 g/mol



Chain-chain entanglements inhibit drawing



Increasing stretch ratio



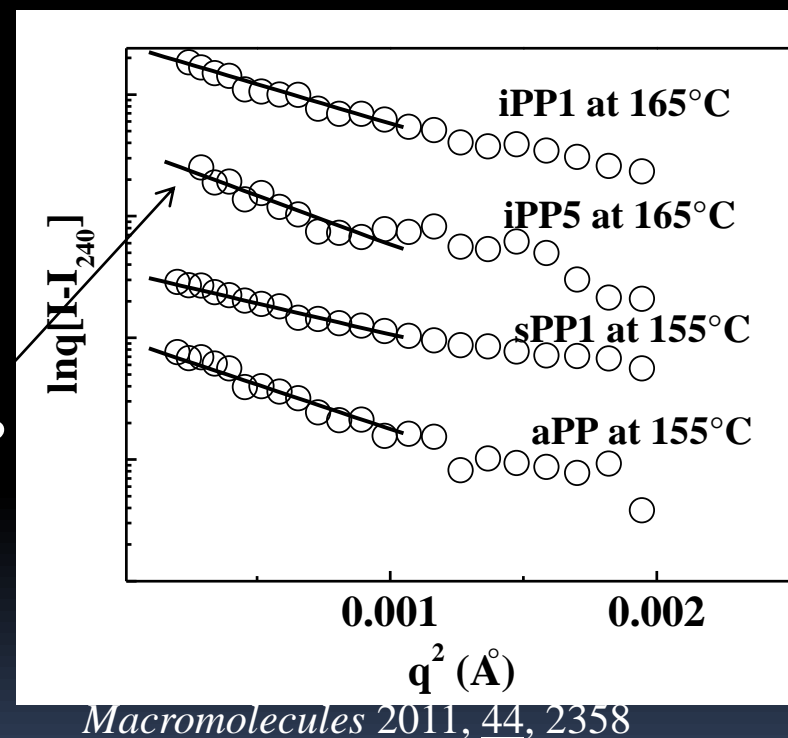
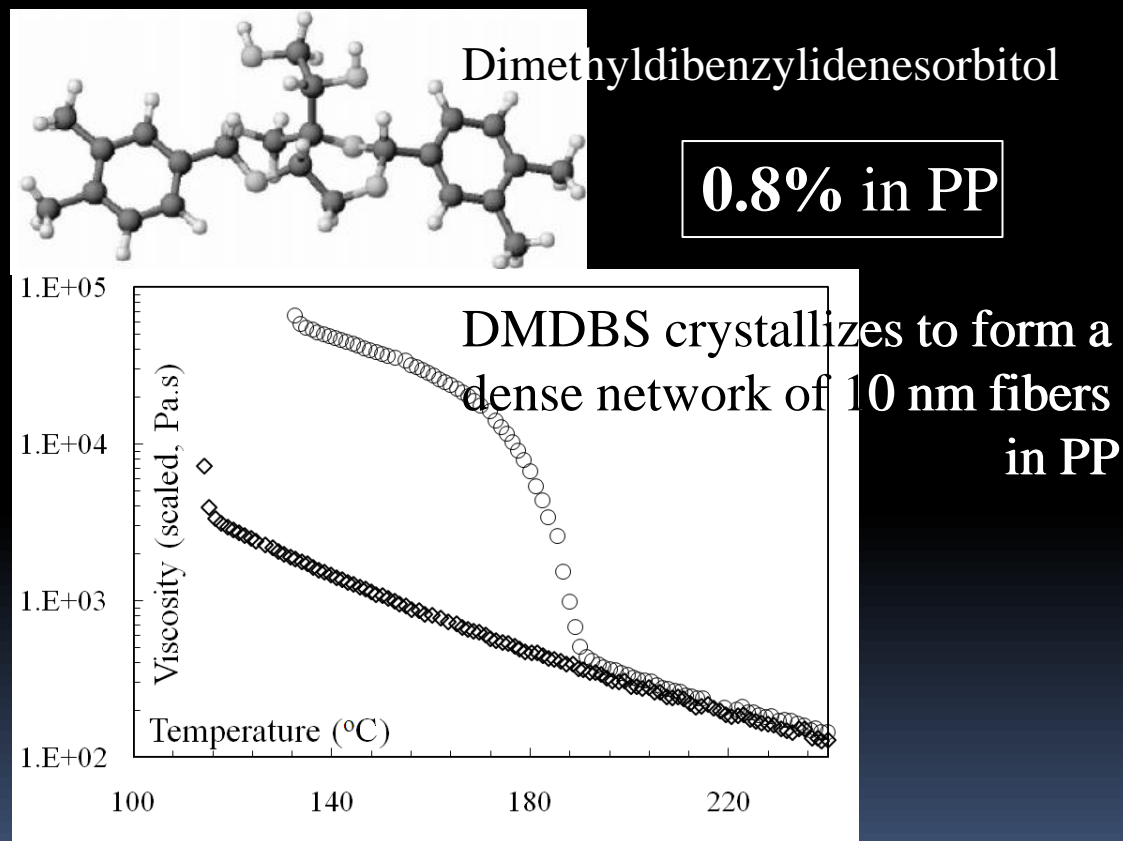
	Tensile Strength [GPa]	Modulus [GPa]
Disentangled UHMWPE	3.75	157
Entangled UHMWPE Grade A	1.95	102
Entangled UHMWPE Grade B	2.30	165
Solution spun Fiber	3.50	135

Macromolecules 2011, 44, 5558



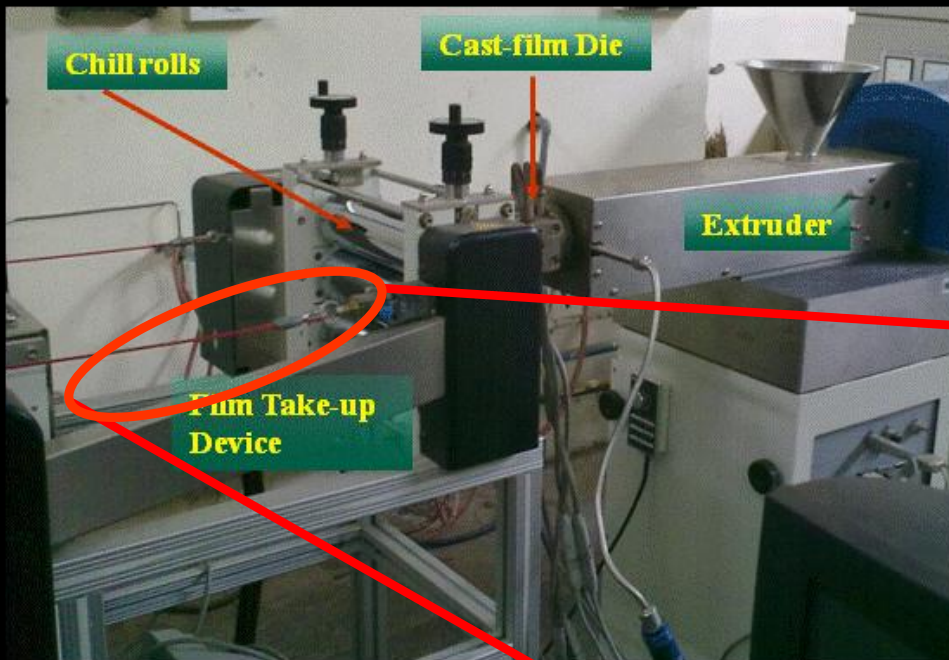
A strategy to induce orientation in PP: Oriented Scaffolds

Generate an oriented scaffold that locally stretches the polymer and forces oriented crystallization

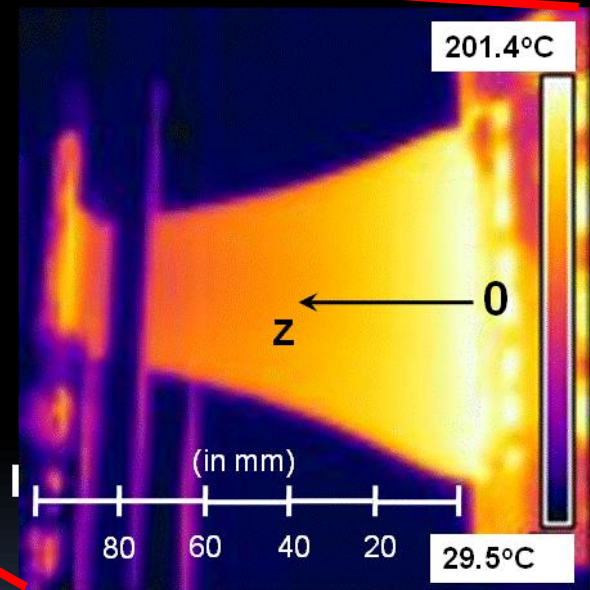
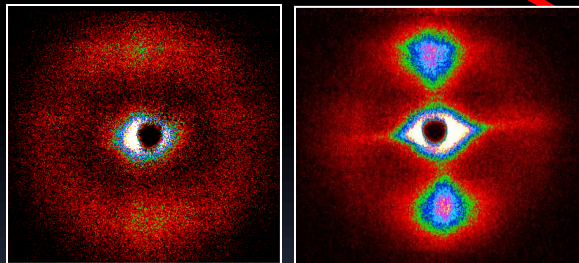




Making oriented films



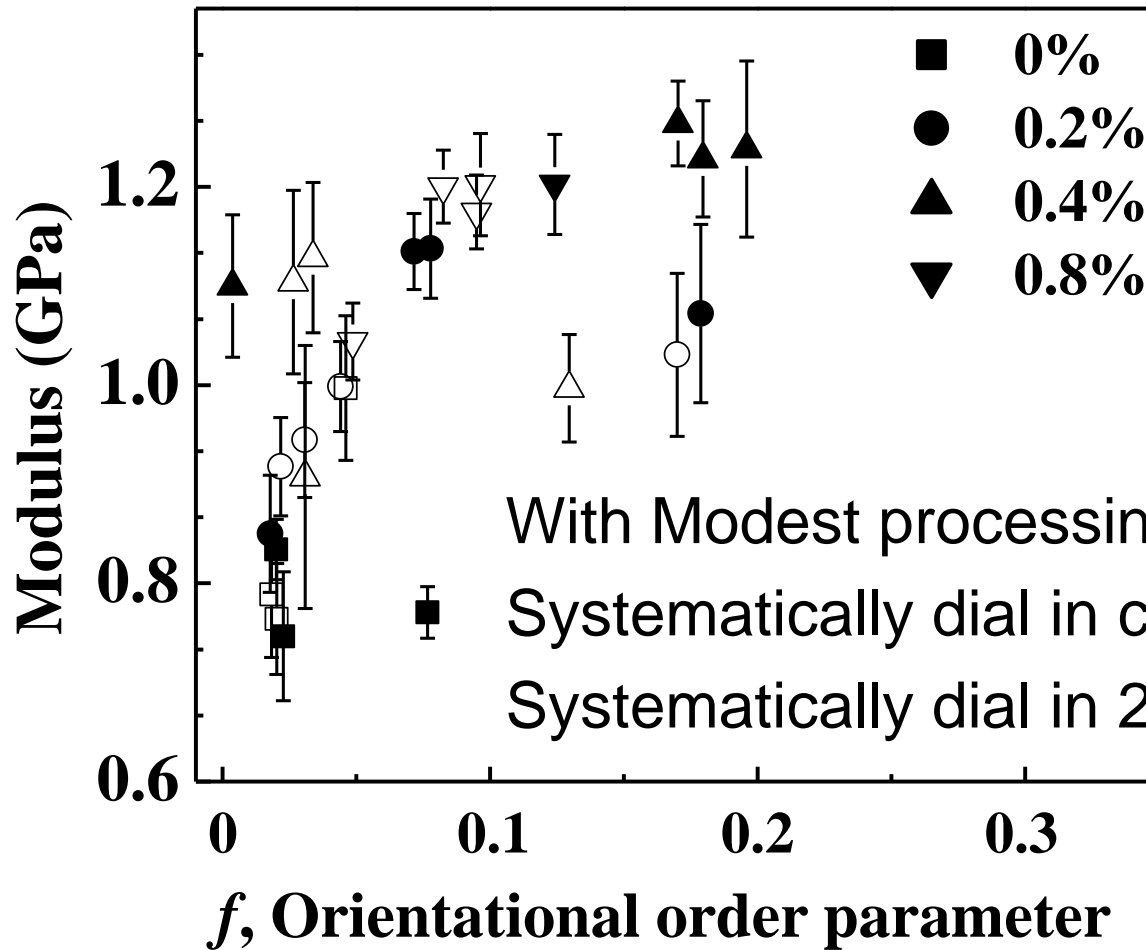
Cool to form DMDBS network
Pull to orient scaffold
PP crystallizes on oriented scaffold



Modest draw ratios
Unable to orient neat PP
PP/0.8% DMDBS strongly oriented



Can Dial-in Orientation (and Modulus)

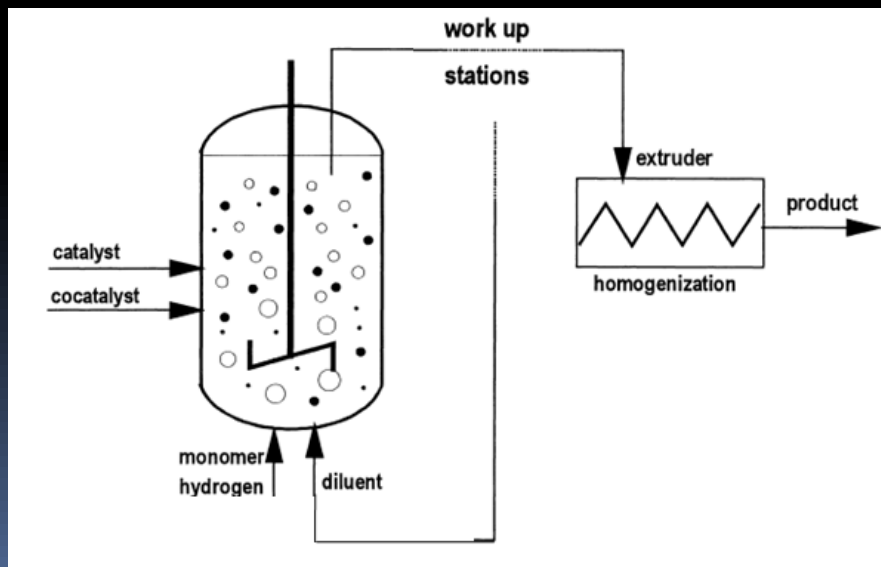


With Modest processing draw ratios, can
Systematically dial in crystal orientation
Systematically dial in 2X modulus variation



Semicrystalline microstructure is complex

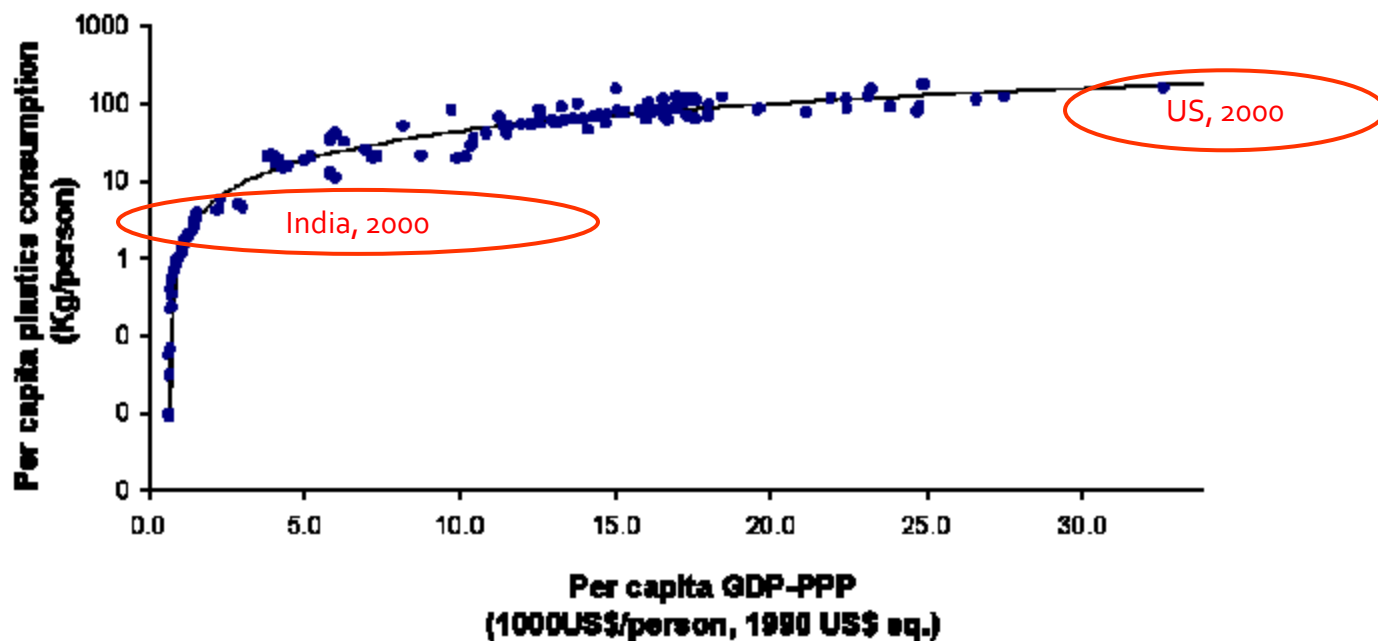
Quantitative measurement of microstructure, and microstructural development, using a combination of experimental tools : insights into connections with properties and with molecular structure



Connecting properties to
microstructure to
molecular structure to
reactor conditions
QUANTITATIVELY



Context: Plastics and Quality of Life



Plastics consumption (y, Kg/person, n = 138) with respect to the GDP in purchasing power Symbols represent data for Australia (1993, 1994, 2000), Austria (1993, 1994), Belgium (1993, 1994, 2000), Chile (1993, 1994), China (1993), Columbia (1993, 1994, 2000), Czech Republic (1993, 2000), Germany (1993, 1994, 2000), Hungary (1993, 1994, 2000), India (1971-2000), Israel (1993, 1994, 2000), Malaysia (2000), Mexico (1993, 1994), New Zealand (1993, 1994), Romania (1993, 1994, 2000), Slovenia (1993, 1994, 2000), South Africa (1990-2000), South Korea (1993, 2000), Spain (1993, 1994, 2000), UK (1993, 1994, 2000), USA (1993, 1994, 2000) and Western Europe (1971-2000). The regression equation is $y = 0.65 \cdot 1.114 x^{0.97}$ (Coefficient of correlation, $R^2 = 0.97$)

Figure 5. Relationship between plastics consumption per capita





Structure Control in Polymers, Polymer Crystallization

“Soft Matter”

Weak interactions ($\sim k_B T$, compare with covalent bond energies of $10^2 k_B T$)

Multitude of interactions

Several examples of out-of-equilibrium systems providing opportunities to *engineer* microstructure and properties

Colloids, Surfactants, Liquid Crystals, Biological fluids, ...

Poster: Using Soap Slime to Assemble Nanoparticles

